

Total Synthesis of (-)-Anisomycin¹⁾

Tatsuya SHONO* and Naoki KISE

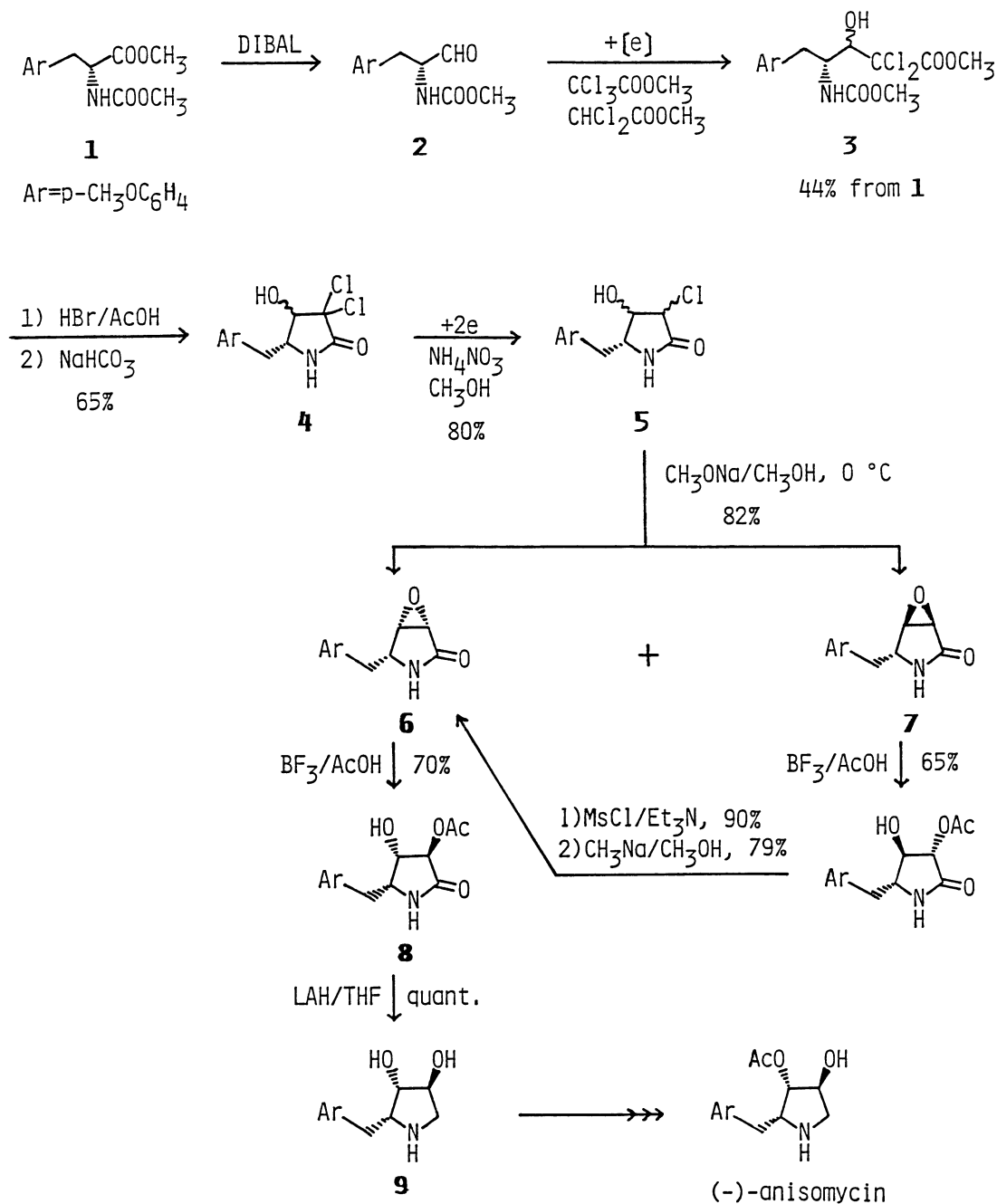
Department of Synthetic Chemistry, Faculty of Engineering,
Kyoto University, Yoshida, Sakyo, Kyoto 606

Antibiotic (-)-anisomycin was synthesized starting from D-tyrosine using the addition of electrogenerated dichloro(methoxycarbonyl)methyl anion to aldehyde 2 as a key reaction.

The antibiotic anisomycin²⁾ is a fermentation product of various *Streptomyces* species and possesses wide activity against pathogenic protozoa and fungi.³⁾ Anisomycin and deacetylanisomycin are used as fungicides to eradicate bean mildew.⁴⁾ The structure of anisomycin has been determined by X-ray investigation⁵⁾ and the absolute configuration by chemical correlation.⁶⁾ Although some methods for the total synthesis of (-)-anisomycin⁷⁾ have already been studied,⁹⁾ these methods required multistage and were not necessarily satisfactory. We wish to report herein the synthesis of (-)-anisomycin from D-tyrosine in which the addition of electrogenerated dichloro(methoxycarbonyl)methyl (DMM) anion to an aldehyde is used as a key reaction (Scheme 1).¹⁰⁾

The optically active aldehyde 2¹¹⁾ was obtained by the reduction of protected D-tyrosine 1 with DIBAL.¹²⁾ The addition reaction of the electrogenerated DMM anion to aldehyde 2 was carried out according to our method.^{13,14)} The adduct 3 was hydrolysed with acid and then treated with sodium bicarbonate to give γ -lactam 4 as a mixture of *cis* and *trans* isomers. Selective reduction of 4 to monochloride 5 was accomplished by cathodic reduction.^{15,16)} Treatment of the isomeric mixture of 5 with sodium methoxide at 0 °C afforded α,β -epoxy- γ -lactams 6¹⁷⁾ and 7 in about 1:1 ratio. After separation of 6 and 7 by column chromatography on

silica gel, **7** could easily be converted to **6**. Thus, treatment of **7** with catalytic amount of $\text{BF}_3\text{-HOAc}$ in acetic acid followed by mesylation and treatment with



Scheme 1.

a base gave **6**. *cis*-Epoxide **6** gave (-)-deacetylanisomycin **9**¹⁸⁾ through treatment with $\text{BF}_3\text{-HOAc}$ and LAH reduction of the resulting acetate **8**.²⁰⁾ Three methods have

already been reported for transformation of 9 to (-)-anisomycin.^{8b,9a,b)}

One of the authors (T.S.) wishes to thank the Ministry of Education, Science, and Culture, Japan, for the Grant-in-Aid for Scientific Research (B) (Grant No. 59470079).

References

- 1) Electroorganic Chemistry. 102.
- 2) B. A. Sobin and F. W. Tanner, Jr., *J. Am. Chem. Soc.*, **76**, 4053 (1954).
- 3) J. E. Lynch, A. R. English, H. Banck, and H. Deligianis, *Antibiot. Chemotherapy*, **4**, 844 (1954); A. Jimenez and D. Vazquez, "Antibiotics," ed by F. E. Hahn, Springer Verlag, Berlin (1979), pp. 1-19.
- 4) "The Merck Index," 10th ed, ed by M. Windholz, Merck, Rahway, New Jersey (1983), p. 98.
- 5) J. P. Schaefer and P. J. Wheatley, *J. Chem. Soc., Chem. Commun.*, **1967**, 578; *J. Org. Chem.*, **33**, 166 (1968).
- 6) C. M. Wong, *Can. J. Chem.*, **46**, 1101 (1968).
- 7) The multistage synthesis of (±)-anisomycin has been reported.⁸⁾
- 8) a) S. Oida and E. Ohki, *Chem. Pharm. Bull.*, **17**, 1405 (1969);
b) D. P. Schumacher and S. S. Hall, *J. Am. Chem. Soc.*, **104**, 6076 (1982).
- 9) a) C. M. Wong, J. Buccini, I. Chang, J. Te Raa, and R. Schwenk, *Can. J. Chem.*, **47**, 2421 (1969);
b) I. Ferner and K. Schnker, *Helv. Chim. Acta*, **53**, 754 (1970);
c) J. P. H. Verheyden, A. C. Richardson, R. S. Bhatt, B. D. Grant, W. L. Fitch, and J. G. Moffatt, *Pure Appl. Chem.*, **50**, 1363 (1978);
d) J. G. Buchaman, K. A. Maclean, H. Paulsen, and R. H. Wightman, *J. Chem. Soc., Chem. Commun.*, **1983**, 486.
- 10) Satisfactory spectroscopic and elemental analyses were obtained for all products.
- 11) Since the aldehyde 2 was readily racemized, it was used without purification.
- 12) G. Stork and E. Nakamura, *J. Am. Chem. Soc.*, **105**, 5510 (1983).

- 13) T. Shono, N. Kise, M. Masuda, and T. Suzumoto, *J. Org. Chem.*, **50**, 2527 (1985).
- 14) The mixture of 2 (10 mmol), methyl trichloroacetate (10 mmol), and methyl dichloroacetate (20 mmol) was dissolved in DMF (60 ml) containing Et₄NOTs (10 g) and electrochemically reduced by using a cell equipped with carbon rod electrodes and a ceramic diaphragm (0.1 A, 20 mF).
- 15) T. Shono, N. Kise, and T. Suzumoto, *J. Am. Chem. Soc.*, **106**, 259 (1984).
- 16) The electroreduction of 4 (5 mmol) was carried out by using a Pt cathode and a carbon rod anode at 0 °C in 0.2 M NH₄NO₃/methanol (40 ml). The cathodic and anodic chambers were divided by a ceramic diaphragm and 3 F/mol of electricity was passed (0.2 A).
- 17) 6: mp 141-142 °C.
- 18) 9: mp 172-174 °C (lit.¹⁹) 176-178 °C); [α]_D²⁰ -20° (c 1.0, ethanol) (lit.¹⁹) -20°); 400 MHz ¹H NMR (CDCl₃-d₆ DMSO) δ 2.64 (1H, dd, *J*=2.5, 12.2 Hz), 2.71 (1H, dd, *J*=7.2, 13.6 Hz), 2.86 (1H, dd, *J*=7.2, 13.6 Hz), 2.98 (3H, br s), 3.26 (1H, dt, *J*=3.4, 7.2 Hz), 3.38 (1H, dd, *J*=5.9, 12.2 Hz), 3.73 (1H, d, *J*=3.4 Hz), 3.78 (3H, s), 4.07 (1H, dd, *J*=2.5, 5.9 Hz), 6.80 (2H, d, *J*=8.7 Hz), 7.20 (2H, d, *J*=8.7 Hz).
- 19) J. J. Beerboom, K. Butler, F. C. Pennington, and I. A. Solomons, *J. Org. Chem.*, **30**, 2334 (1965).
- 20) 8: mp 175-176 °C; [α]_D²⁰ +6.2° (c 0.5, ethanol).

(Received January 31, 1987)